

## Determination of $^{236}\text{U}$ in a Th target irradiated with Li ions by ICP mass spectrometry

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Neptunium-237 ( $T_{1/2} = 2.1 \times 10^6$  y) released from nuclear facilities is an important radioactive target nuclide that contaminates the environment.<sup>1)</sup> The nuclide can be analyzed by accelerator mass spectrometry; however, a spike material of neptunium for the measurement has not been obtained, thus far. We previously proposed the production of neptunium-236 ( $T_{1/2} = 1.5 \times 10^5$  y) in a Th + Li reaction as a candidate for the material.<sup>2,3)</sup> However, the material prepared from the irradiated target may be contaminated with long-lived isobars of  $^{236}\text{Pu}$  and  $^{236}\text{U}$ , which are the decay products from  $^{236\text{m}}\text{Np}$ , or a byproduct in the reaction. In addition,  $^{235}\text{U}$  contained in Th as an impurity, may also absorb neutrons and produce  $^{236}\text{U}$ . In this study, we aimed to determine U isotopes contained originally in the Th target as well as those produced in the nuclear reaction by High Resolution ICP-MS (HR-ICP-MS). The amount of  $^{236}\text{U}$  produced in the reaction can be estimated by measuring a neutron flux during irradiation through the use of a monitoring foil of Au.

We irradiated a  $^{232}\text{Th}$  target with 43 MeV  $^7\text{Li}$  ions at the RIKEN AVF cyclotron. Au metal foils (ca. 200 mg each) and electrodeposited U samples (ca. 1 mg each) were placed at the end of the beam course to function as a neutron monitor and a reference, respectively, as illustrated in Fig. 1.

After irradiation, the neutron flux was obtained from activation of the Au foil measured by  $\gamma$ -ray spectroscopy using a Ge semiconductor detector. The U-electrodeposited sample was dissolved in a concentrated

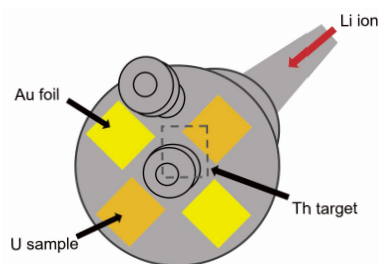


Fig. 1. Experimental setup of neutron monitor foils (Au) and U targets attached to the end of a target holder of an irradiation beam course at RIKEN.

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Table 1. U isotopic ratios in U-electrodeposited samples by ICP-MS.

	Irradiated		Unirradiated
	U-1	U2	U-3
	Ratio%		Ratio%
235/238	0.62~0.82	0.61~0.84	0.63~0.85
236/238	0.5×0.82 <sup>-4</sup>	(1-7)×0.82 <sup>-4</sup>	N.D

nitrate solution. Subsequently, the solution was diluted so that the U concentration was 0.1 or 1 ng/g, as measured by HR-ICP-MS with a reference solution of XSTC-13 supplied by SPEX Industries, Inc. for calibration.

From the measurements of photopeak intensities due to  $^{198}\text{Au}$  and  $^{196}\text{Au}$  in the Au foils, the thermal neutron flux and fast neutron flux were  $6.5 \times 10^5 \text{ s}^{-1}$  and  $1.6 \times 10^7 \text{ s}^{-1}$ , respectively. These values were calculated from induced radioactivities and the cross section data<sup>4)</sup> of  $(n, \gamma)$  and  $(n, 2n)$  reactions. The isotope measurement results of the U-electrodeposited sample by HR-ICP-MS are shown in Table 1, where the ratios denote the range of observation due to the large dispersion of the data depending on the sample. The abundance ratios of  $^{235}\text{U}$  to  $^{238}\text{U}$  for all the samples are approximately 0.7%, corresponding to the natural abundance of U. Some events due to  $^{236}\text{U}$  were observed for the samples attached to the target holder. However, the events due to the unirradiated sample was less than the detection limit. It was confirmed that  $^{236}\text{U}$  was produced in the U-electrodeposited samples. The amount of  $^{236}\text{U}$  produced agrees with the amount estimated from the thermal neutron flux by the monitor within some uncertainty of the data. Based on the measured concentration of U and the measured thermal flux, the number of  $^{236}\text{U}$  atoms produced in the target sample was  $(0.9 - 2.0) \times 10^2$ , much less than the number of  $^{236\text{g}}\text{Np}$  atoms estimated from the measured cross section,<sup>2)</sup> that is, approximately 1 mb. Thus, the amount of U was concluded to be negligible to contaminate a Np tracer prepared from the irradiated target.

### References

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