

Adsorption behavior of No with a TTA chelate extractant from HF/HNO₃ acidic solutions

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In a study on the fluoride complexation of a super-heavy element, rutherfordium (Rf), we focus on the extraction with an acidic chelate extractant, 2-thenoyltrifluoroacetone (TTA), which is sensitive to the valence of the metal complex. Recently, we developed a reversed-phase chromatography technique with TTA and performed Rf experiments with this technique in various HF/0.01 M HNO₃ concentrations.¹⁾

The isotope ²⁶¹Rf used in the experiments decays into its daughter ²⁵⁷No. The α -particle energies of ²⁵⁷No ($E_{\alpha} = 8.22, 8.32$ MeV) are close to that of ²⁶¹Rf ($E_{\alpha} = 8.28$ MeV). Therefore, these energies are difficult to distinguish from each other. In the Rf experiments, two types of ²⁵⁷No α -events are expected to be observed. One is from ²⁵⁷No produced from the α -decay of ²⁶¹Rf after its chemical separation. It reflects the chemical behavior of Rf. The other is of ²⁵⁷No deposited during the collection of ²⁶¹Rf, which reflects the chemical behavior of No. In order to correct the contribution of ²⁵⁷No, we observe the adsorption behavior of No in the same systems of the Rf experiments.

Similar to the Rf experiments, the isotope ²⁵⁵No ($T_{1/2} = 3.10$ min) was produced in the ²⁴⁸Cm(¹²C, 5n) reaction with an 84 MeV ¹²C beam at the RIKEN K70 AVF cyclotron. The reaction products were rapidly transported with a KCl/He gas-jet system to a chemistry laboratory and were deposited on the collection site of the on-line Automated Rapid Chemistry Apparatus (ARCA) for 180 s. After deposition, the products were dissolved in 85 μ L of various HF/0.01 M HNO₃ solutions and fed into a 1.6 mm i.d. \times 7.0 mm TTA resin column at a flow rate of 0.1 mL/min. The effluent from the column was collected on a Ta disk as fraction 1. The remaining products in the column were then stripped with 250 μ L of 0.1 M HF/0.1 M HNO₃ solution at a flow rate of 1.0 mL/min and then collected on another Ta disk as fraction 2. Both samples were evaporated to dryness using hot He gas and a halogen heating lamp. The samples were assayed with a rapid α /SF detection system for the aqueous chemistry of super-heavy elements at RIKEN. In order to determine the chemical yield, ¹⁶²Yb was simultaneously produced from the Gd content in the Cm target and was measured by a Ge detector after the measurement of ²⁵⁵No. The average chemical yield of ¹⁶²Yb

in all the experiments was approximately 17%.

From 195 cycles of the No experiments, a total of 1042 α -events indicating the production of ²⁵⁵No were registered in the energy range of 7.60–8.20 MeV. The adsorption probability, %ads, of No with a fixed volume of the effluent was evaluated using the following equation:

$$\%ads = Fr2 / (Fr1 + Fr2) \times 100, \quad (1)$$

where Fr1 and Fr2 are the radioactivities observed in fraction 1 and 2, respectively. The decay of No was taken into account in the correction for the %ads values. The results for %ads of ²⁵⁵No as a function of $[F^-]_{eq}$ in the range of 1.93×10^{-5} to 1.66×10^{-3} M are shown in Fig. 1. In the Rf experiments,¹⁾ the %ads values of ²⁶¹Rf were constant at approximately 60% in the $[F^-]_{eq}$ range up to 5×10^{-4} M and then steeply decreased at $[F^-]_{eq} = 9 \times 10^{-4}$ M. On the other hand, in the No experiments, it was found that the %ads values of ²⁵⁵No were less than 10% in the entire range of $[F^-]_{eq}$.

In order to evaluate the %ads values of ²⁶¹Rf, we assumed in the previous report¹⁾ that the adsorption of No was negligible. Based on the present work, it was confirmed that No was adsorbed to TTA to a small extent and the %ads values of ²⁶¹Rf can be determined with greater precision.

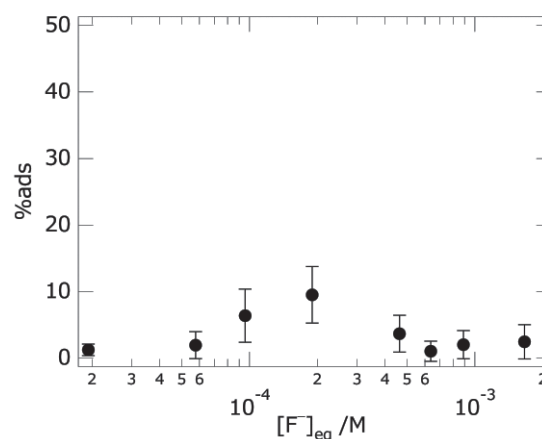


Fig. 1 Adsorption probability, %ads, of ²⁵⁵No plotted as a function of $[F^-]_{eq}$ in TTA column chromatography.

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Reference

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